

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

31093. LEVITINA, R. F.

Fizicheskoe razvitiye novorozhdennykh nedonoshennyykh detey v Leningrade  
v 1945g. Voprosy pediatrii i okhrany misterinstva i detstva, 1949, vyp. 4,  
s. 17-23

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

LEVITINA, S.

First meeting of young inventors and innovators from Moscow enter-  
prises. Izobr. v SSSR 3 no. 2:45-46 p '58.  
(Moscow--Inventors) (MIRA 11:3)

LEVITINA, S.

Valuable manuals on industrial safety. Okhr.truda i sots.  
strakh. no.2:93 Fe '59. (MIRA 12:4)  
(Industrial safety)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITINA, S.M., inzh.

School for efficiency promoters. Izobr. i rats. no.6:11-15  
Je '58. (MIRA 11:9)

(Technical education)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITINA, S.M., trubet USHATKIN, I.P., inach.

New equipment and advanced technological processes in the  
spinning industry. Mekt. 1 avt.prizv. 18 no.8:54-57 Ag 164.  
(MIRA 17:10)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITINA, S.M., inzh.

Mechanization and automation in the textile industry. Mekh. i  
avtom. proizv. 19 no.5:53-55 My '65. (MIRA 18:11)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

LEVITINA, S.M., inzh.

Conferences of readers of the periodical in 1959. Mekh.i avtom.  
proizv. 14 no.3:48-49 Mr '60. (MIRA 13:6)  
(Engineering--Periodicals)

Doc Med Sci

LEVITINA, YE. A.

Dissertation: "Data on the Problem of the Influence of Nerve Centers on the Functional State of a Motive Apparatus."  
26/5/50

Acad Med Sci USSR

SO Vecheryaya Moskva

Sum 71

SO Vecheryaya Moskva

Sum 71

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITINA, Y.E.I.

Behavior of magnesium in alkali solutions. Zhur. ob. khim. 24 no.2:  
216-218 F 154. (MLRA 7:4)  
(Magnesium)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

LEVITINA, Z. I.

USSR/Chemistry - Physical chemistry

Card 1/1 Pub. 151 - 1/37

Authors : Samartsev, A. G., and Levitina, Z. I.

Title : Behavior of magnesium in potassium bichromate solutions

Periodical : Zhur. ob. khim. 24/10, 1697-1700, Oct 1954

Abstract : The rate of development of protective films on magnesium, when subjected to the effect of potassium bichromate solutions containing certain activating additions, was investigated. The passivation of Mg in  $K_2Cr_2O_7$  solutions was established at room temperature and at boiling temperature. Passivity was found to be the result of the formation of a thin non-visible protective film consisting of hardly soluble components. The effect of activating salts on the passivation of the metal and the relation between the activating salts and the temperature and acidity of the solution, are explained. Four references: 2-USSR; 1-German and 1-USA (1907-1954). Graphs.

Institution : ...

Submitted : May 13, 1954

LEVITINOV, S.D., dotsent; POLYAKOV, G.V., inzh.; ASTRAKHANTSEV, N.Ya.,  
inzh.; POGORELOV, G.M., inzh.

Recuperative braking on commercial electric locomotives in open-  
pit mines. Izv. vys. ucheb. zav.; gor. zhur. 6 no.4:122-135 '63.  
(MIRA 16:7)

1. Chelyabinskij politekhnicheskiy institut. Rekomendovana  
kafedroy elektroprivoda i avtomatizatsii promyshlenniykh ustanovok.  
(Mine railroads--Brakes)

LEVITMAN, Kh. Ya.

YARMOLENKA, N.F., professor; LEVITMAN, Kh.Ya., kandydat tekhnicheskikh naук.

High molecular compounds in solutions of salt mixtures based on alkalinity data. Vestsi AN BSSR no.1:67-66 Ja-P '52. (MLRA 7:8)

1. Pravadezyny chlen AN BSSR (for Yarmolenka)  
(Systems (Chemistry)) (Molecular dynamics)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

*LEVITMAN A. A.*

USSR/Analysis of Inorganic Substances

G-2

Abs Jour: Ref Zhur-Khimiya, No 6, 1957, 19648

Author : Kh. Ya. Levitman, Z. A. Krivchik

Inst :

Title : Amperometric Determination of Copper and Nickel  
in Alloyed Steel with Ruboanehydride Acid.

Orig Pub: Zavod. Laboratoriya, 1955, 21, No 4, 397 - 399.

Abstract: The method is based on the difference of the solubility of Cu and Ni rubenates, which renders it possible to determine both these metals according to the break in the left hand branch of the V-shaped titration curve during the same titration process. Titration was carried out at 1.3 v on the mercury drop electrode in an ammonium

Card 1/2

- 126 -

PLATINUM PLATE

Phototurbidimetric determination of copper in baths for  
nickel plating. Kh. V. Leont'ev and N. M. Vladyskina.  
Reform. Politekhnicheskii Institut, Sverdlovsk, USSR.  
Radio 1955, Vol. 17, No. 107. Absorbance titration Cu from cal-  
ibration curves and by titration were studied with a dif-  
ferential photometer by using the sol formed with  
rubeanic acid. The Ni-plating bath contained  $\text{NaSO}_4$ ,  
 $\text{H}_2\text{O}$  175,  $\text{Na}_2\text{SO}_4$  101.0,  $\text{NiCl}_2$  29, and  $\text{H}_3\text{BO}_3$  20 g.  
20, 30, 40, 60, 70, 90, 110, 130, and 150 mg./l. Cu as  $\text{CuCl}_2$   
was present. The ability of an 8% soln. of rubeanic acid  
formed an olive-green sol in the presence of gelatin. Peaks  
in the absorption curve (at 400, 420, 440 mg./l. of Cu) were at  
 $\lambda = 420$  and 7000 Å. Vials (1000, 1700 Å.), red (1000  
7500 Å.) and stroke-colored glass filters were compared.  
The soln. to be analyzed was mixed from (in order) the  
plating bath 2, water, 90, 0.5% gelatin soln., 2, 80% AcOH 1,  
and 8% soln. of rubeanic acid (80 mg./l.) 3 ml. The frac-  
tion of light absorbed was linear, ranging from 0.086 (20 mg.  
Cu/l. of bath) to 0.517 (150 mg. Cu/l. of bath) for the violet  
filter; 0.036 (20 mg. Cu/l.) to 0.310 (150 mg. Cu/l.) for the  
red filter, and 0.027 (20 mg. Cu/l.) to 0.237 (150 mg.  
Cu/l.) for the smoke-colored glass. Four samples contg  
50, 80, 100, and 120 mg./l. of Cu were analyzed with errors  
of 6, 0, 3, and 1.7%, resp. The soln. was titrated (in order of  
mixing) plating bath, 0.5% gelatin soln., 2, 80% AcOH  
0.5 ml., with water to 50 ml. Rubeanic acid soln. was  
added from a microburet, and the light absorption was  
measured. In 4 expts. with solns. contg. 0.15, 0.20, 0.29,  
and 0.25 mg. Cu/l., the errors were 6.6, 0.5, 2.0, and 8.0%,  
resp. Fe, Pb, and Zn did not interfere in the presence of  
AcOH.

R. D. Mich

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

USSR/Thermodynamics. Thermochemistry. Equilibria. Physico-Chemical B-8  
Analysis. Phase Transitions

Abs Jour : Ref Zhur - Khimiya, No 8, 1957, 26148

Author : N.F. Yermolenko, Kh.Ya. Levitman

Title : Study of Molecular Compounds in Solutions by Method of Physico-Chemical Analysis by Measurement of Refraction Index.

Orig Pub : Zh. neorgan. Khimii, 1956, 1, No 6, 1162-1172

Abstract : A series of ternary systems of aqueous and alcoholic solutions of mixed mineral salts was studied by measuring the index of refraction ( $n$ ) and by investigating the deviations from additivity by curves  $\Delta n$ -composition. The possibility of the application of refraction measuring to the detection of molecular compounds in a solution is shown.

Card : 1/1

LEVITMAN, KH. YA.

137-58-1-2134

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 1, p 290 (USSR)

AUTHORS: Levitman, Kh. Ya., Yermolenko, N. F. [ Levitman, Kh. Ya.,  
Yermolenko, N. F. ]

TITLE: Amperometric Analysis of Copper in Electrolytic Baths for  
Nickel, Zinc and Cadmium Plating (Amperometricheskoye  
opredeleniye medi v gal'vanicheskikh vannakh dlya nikelirovaniya,  
tsinkovaniya i kadmirovaniya) [ In Belorussian ]

PERIODICAL: Vestsi AN BSSR, Ser. fiz.-tekhn. n., Izv. AN BSSR, Ser.  
fiz.-tekhn. n., 1956, Nr 4, pp 133-137

ABSTRACT: A description is offered of a method of analyzing for small  
quantities of Cu in electrolyte baths (EB) by amperometric  
titration employing rubeane; the advantages of the employment  
thereof as a precipitant are indicated. The titration was per-  
formed in a visual polarimeter. The content of Cu in EB for  
nickel, zinc, and cadmium plating was established by the  
readings of a mirror galvanometer sensitive to  $2.16 \times 10^{-9}$  amps  
at 1 mm/m. In preparation, the initial solutions simulating the

Card 1/2

137-58-1-2134

**Amperometric Analysis of Copper (cont.)**

compositions of the EB were made of mixtures of the respective pure salts, and optimum voltages were used in the amperometric titration for Cu. The amount of Cu in real EB for nickel, zinc, and cadmium plating was then determined at concentrations of 20-500 mg/l in the bath mixtures. The method developed makes possible determination of Cu in an acetic acid medium in the presence of added Fe, Zn, and Pb without preliminary separation thereof.

P.P.

- 1. Copper—Determination    2. Electrolytic titration—Equipment    3. Electrolytic cells—Equipment**

Card 2/2

Levitman, Kh. Ya.

USSR/ Analytical Chemistry - Analysis of Inorganic Substances

G-2

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 12039

Author : Tananayev I.V., Levitman Kh.Ya.

Inst : Commission on Analytical Chemistry of the Academy of Sciences USSR

Title : Photo-Turbidimetric Determination of Copper in Aluminum Alloys by Means of Rubeanic Acid

Orig Pub : Tr. Komis. po analit. khimii AN SSSR, 1956, 7(10), 21-26

Abstract : Photo-turbidimetric determination of Cu in solution of aluminum alloys containing Fe and Mg, is effected by titration with a standard solution of rubeanic acid, in an acetic acid medium. Accuracy 5-6% (relative).

Card 1/1

YERMOLENKO, N.P.; LEVITMAN, Kh.Ya.; ZARUBKINA, A.K.

Effect of concentration of molecular compounds in mixed salt  
solutions on the stability of these compounds. Uch.zap. BGU  
no.29:251-256 '56. (MIRA 11:11)  
(Solution (Chemistry))

ALL INFORMATION CONTAINED  
HEREIN IS UNCLASSIFIED

USSR/Physical Chemistry - Electrochemistry

B-12

Abs Jour : Referat Zhur - Khimiya, No 2, 1957, 3976

Author : Levitman Kh.Ya., Karpovich N.I., Rutskaya Ye.I.

Inst : Belorussian Polytechnic Institute

Title : Polarographic Properties of Rubeanic Acid

Orig Pub : Sb. nauch. rabot Belorus. politekhn. in-ta, 1956, № 55,  
112-118

Abstract : Rubeanic acid (I) undergoes reduction in acid, neutral and alkaline solutions At pH 2-6 I gives one clearly defined wave. On increase of pH height of the wave (h) increases and E<sub>1</sub> becomes more negative. At pH 7-11.4 I produces three waves. In acid solutions h is proportional to concentration of I. In neutral and alkaline media h of second and third wave is also proportional to concentration of I; the second wave is convenient for a determination of I.

Card 1/1

- 216 -

LEVITMAN, Kh.Ya.; KRYUCHYK, Z.A.

Use of magnesium chloride as a background for polarography.  
Vestsi AN BSSR. Ser. fiz.-tekhn.nauk. no.2:43-47 '58.  
(MIRA 11:10)  
(Polarography) (Magnesium chloride)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITMAN, Kh.Ya.; VLADYKINA, N.M.

Photocolorimetric determination of the activity of charcoal.  
Sprint. prom. 24 no.8:28-29 '58. (MIRA 11:12)  
(Carbon, Activated) (Colorimetry)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

LEVITMAN, Kh.Ya., kand.tekhn.nauk; GORSKAYA, Ye.V.

Physicochemical analysis of the system silver nitrate-rubecanic  
acid-water. Sbor.nauch.rab.Bel.politekh.inst. no.63:164-170  
'58. (MIEA 12:4)

(Silver nitrate) (Oxamide)

YERMOLENKO, N.F. [Iarmolenko, N.F.]; DEYCH, A.Ya. [Deich, A.IA];  
LEVITMAN, Kh.Ya. [Levitman, Kh.IA]

Molecular compounds in ternary and binary mixtures based on  
refraction and density factors. Vestsi AN BSSR. Ser. fiz.-tekhn.  
nav. no.1:25-29 '59. (MIR 12:6)  
(Systems (Chemical))

LEVITMAN, Kh.Ya., kand.tekhn.nauk

Methodological instructions for laboratory work in the physical  
chemistry course. Shor. metod. rah. Bel. politekh. inst. no. 1:53-  
56 '59. (MIRA 14:1)  
(Chemistry, Physical and theoretical--Study and teaching)

LEVITMAN, Kh.Ya., kand.tekhn.nauk

Remarks concerning methods of work of an assistant in the  
department of chemistry. Sbor. metod. rab. Bel. politekh. inst.  
no. 1:129-134 '59. (MIRA 14:1)  
(Chemistry--Study and teaching)

LEVITMAN, Kh.Ya.; RUTSKAYA, Ye.I.; KARPOVICH, N.I.

Physicochemical analysis of a lead nitrate - rubeanic acid - water  
system and its importance in analysis. Sbor.nauch.trud.Bel.politekh.  
inst. no.87:45-54 '69. (MIRA 14:4)  
(Lead nitrate) (Oxamide)

LEVITMAN, Kh.Ya.; GORSKAYA, Ye.V.

Turbidimetric determination of silver using rubeanic acid.  
Sbor.nauch.trud.Bel.politekh.inst. no.87:55-65 '59. (MIRA 14:4)  
(Oxamides) (Silver—Analysis)

LEVITMAN, M.Kh., vrach-rentgenolog

Congenital diaphragmatic hernia. Zdrav. Bel. 7 no. 2:54-55 P '61.  
(MIRA 14:2)

1. Iz IV klinicheskoy bol'nitsy g.Minska (glavnnyy vrach Ye.M.  
Sel'dimirova).  
(HERNIA)

POPOV, Viktorin; LEVITINA, S.A., red.; PINKHASOV, Ya.V.

[Chardzhou-Kungrad; along the track with a notebook]  
Chardzhou - Kungrad; s bloknotom po trasse. Tashkent,  
Gos.izd-vo UzSSR, 1947. 29 p. (MIRA 16:8)  
(Soviet Central Asia--Railroads--Location)

LEVITINA, S.M., inzh.

From pages of the periodical "Mekhanizatsiya i avtomatizatsiya  
proizvodstva." Tekst.prom. 21 no.5:89 My '61. (MIRA 15:1)  
(Textile industry)

LEVITMAN, S. YA.

FA 30/47 16

USSR/Chemistry - Systems  
Chemistry - Acetic Acid

Sep 48

"Change in the Refractive Index in the System  
 $\text{CH}_3\text{COOH}-\text{H}_2\text{O}$ ," S. Ya. Levitman, N. F. Yermolenko,  
Inst. Chem., Acad. Sci. Belorussian SSR, 52 pp

"Zhur Obshch Khimi" Vol XVII, No 9

Studies refractive index of acetic acid-water  
mixtures of composition 10,20, and 30%. Es-  
tablishes existence of molecular compound  
 $\text{CH}_3\text{COOH} \cdot 2\text{H}_2\text{O}$  by deviation from additivity.  
Establishes existence of molecular compound  
 $\text{CH}_3\text{COOK} \cdot \text{H}_2\text{O}$  by composition maximum. Latter

30/472

30/472

USSR/Chemistry - Systems (Contd)

Sep 48

Type molecular compound is most usual for acetic acid.  
Submitted 4 Mar 48.

1  
"Physicochemical analysis of systems significant in analytical chemistry. XVII. Study of the reaction of copper dithiophosphate formation with the aid of light absorption. I. V. Taranov and G. Ya. Levitan (Bulg.)

Russian Polytech. Inst., Minsk). Zash. naub. Khim. 4, 212-19(1949); cf. C.A. 44, 4744. -Extinction readings of solutions containing different proportions of Cu and the oxamide indicate that the ratio Cu:pept. remains, as has been assumed, one mole Cu to one mole dithiophosphate. By means of the reaction,  $Cu^{++}$  can be titrated with an  $\text{Hg}^{++}$  soln. of the reagent or the turbidity produced with very small quantities of Cu can be measured and compared with that produced with known quantities of Cu.

M. Hoveh

(A)

Higher-molecular compounds in solutions of salt mixtures, according to refractometric data. N. V. Krasil'ko and B. Ya. Levitan, Zhur. Osnovn. Khim., (J. Gen. Chem.) 26, 3177 (1953). Formation of a compd. between 2 salts in aq. soln. is indicated by a max. on the curve representing the deviation  $\Delta n$  of the actual refractive index  $n_0$  of the mixt. from the theoretical  $n_{calcd}$ . from the index of the pure simple solns. by the additivity rule, plotted in terms of the mol. concns. of one component, with the sum of the mol. concns. of the 2 salts kept const. For the system LiCl-NaCl-H<sub>2</sub>O at 20°,  $\Delta n$  is small at all LiCl:NaCl ratios, and varies irregularly; consequently, no compnd. is formed in that system. In MgCl<sub>2</sub>-KCl-H<sub>2</sub>O, in BaCl<sub>2</sub>-KCl-H<sub>2</sub>O, and in NH<sub>4</sub>Br-NH<sub>4</sub>Cl-H<sub>2</sub>O,  $\Delta n$  passes through a distinct max. at the mol 1:1 ratio of the 2 salts. Consequently, there is an indication of the compds. K[MgCl<sub>2</sub>], NH<sub>4</sub>[Sc(+)<sub>2</sub>], and K[BaCl<sub>2</sub>]. N. Todor

LEVITMAN, V.S.; KUSURINA, N.N.; ROSIKOVA, T.N.

[Program for calculating many-storied multispan frames  
using the BESM-2M digital computer; program SIDR-3] Program-  
ma rascheta mnogotazhnnykh mnogoproletnykh ram na elektronnoi  
mashine BESM-2M; programma SIDR-3. Moskva, 1964. 234 p.  
(MIRA 18:2)  
(Serija 11-49)

1. Moscow. Gosudarstvennyy institut tipovogo i eksperimental'-  
nogo proyektirovaniya i tekhnicheskikh issledovaniy.

DUPAL, Jaroslav [Dupal, Jaroslav]; GAVLICHEK, Jaromir [Havlicek, Jaromir]; STOCHEK, Ferdinand [Stoces, Ferdinand]; BARTUNEK, Iosif [Bartunek, Josef]; LEVITMAN, Ye.A. [translator]; TULUPNIKOV, A.I., red.; SUMNIK, Z.A., red.; IL'YUSHENKOVA, T.P., tekhn. red.

[Problems in determining the effectiveness of agricultural production in Czechoslovakia] Voprosy opredeleniya effektivnosti sel'skokhoziaistvennogo proizvodstva v Chekhoslovakii. Pod red. A.I.Tulupnikova. Moskva, Gosstatizdat, 1962. 178 p. Translated from the Czech. (MIRA 15:11)

1. Nauchno-issledovatel'skiy institut narodnokhozyaystvennogo planirovaniya pri Gosudarstvennoy planovoy komissii, Chekhoslovakija (for Dupal, Gavlichek). 2. Gosudarstvennaya planovaya komissiya, Chekhoslovakija (for Bartunek).

(Czechoslovakia--Agriculture--Economic aspects)

SEARCHED INDEXED  
SERIALIZED FILED  
B-147B-37

AUTHOR: Levitson A. M.

TITLE: Exchange of experience

PERIODICAL: Zavodskaya laboratoriya po radioelementam i radioisotopam

TEXT: The author developed a simple particle separator for obtaining hydrocarbon gases and carbon dioxide of high purity for use in gas chromatographic instruments. The apparatus is a "integrator" consisting of two described earlier by the author in "Zhurnal Radiofizika i Radiohemmica" (Journal of Radioelectronics and Radiochemistry) and published in "Gazochromatografiya" (Gas Chromatography). The apparatus consists of an apparatus for dehydrating alcohol and one for removing gaseous heat from the apparatus formed in the fermenter. The apparatus consists of two parts. Dehydration of methyl alcohol over aluminum oxide and removal of heat produced by fermentation. Ethanol is converted to methanol in the apparatus. Hydrogen is added to the apparatus. It is used to dilute the alcohol. Air is also admitted to the apparatus. The apparatus is connected to a gas chromatograph. Carbon dioxide is removed from the apparatus.

Exchange of experience.

SAC / JAS / JAS  
P-10 2001

Item is being processed. Equipment is now available for use by the Bureau.  
Complete transmission to be made later.

**ASSOCIATION:** Very briefly mentioned in the first draft of the report.  
The association is complete. A full description will be provided in the final report. It is anticipated that the final report will be completed by the end of the month.

Card 1.

REF ID: A6510

USSR/Mathematics - Eigenfunctions

Card 1/1 : Pub. 22 - 4/44

Authors : Leviton, B. M.

Title : On expansion of the equation  $\Delta u + \{ \lambda - q(x_1, x_2, \dots, x_n) \} u = 0$  by eigenfunctions

Periodical : Dok. AN SSSR 97/6, 961-964, Aug 21, 1954

Abstract : A series of theorems dealing with an expansion of the given equation:

$\Delta u + \{ \lambda - q(x_1, x_2, \dots, x_n) \} u = 0$  by eigenfunctions are considered with provision that  $\frac{\partial u}{\partial n}|_{\partial B} = 0$  is a boundary condition. The theorems

are intended to prove, and do prove the possibility of such expansions for a limited Euclidean space as well as for any unlimited one. Three references: (1936-1954)

Institution : ...

Presented by : Academician S. L. Sobolev, May 25, 1954

USSR/Mathematics

Card 1/1 Pub. 22 - 4/45

Authors : Leviton, B. M.

Title : On the expansion along the eigen functions of the Schrödinger operator in  
the case of an indefinitely growing potential

Periodical : Dok. AN SSSR 103/2, 191-194, Jul 11, 1955

Abstract : A study of the expansion (along the eigen functions of the equation:

$$\Delta u + \{ \lambda - q(x_1, x_2, x_3) \} u = 0$$

of functions which are growing at infinity as a polynomial. The equation  
describes a 3-dimensional space ( $E_3$ ) in which the  $q(x)$  is a continuous real  
function, and the  $x$  is a point in the  $E_3$ . The method used in the study is  
good for any dimensional space, however, for the sake of simplicity, it is  
chosen as 3-dimensional. Six references: 1 USA, 2 Brit. and 3 USSR(1951-1954)

Institution : Artillery Acad. imen F. E. Dzerzhinsky

Presented by : Academician S. L. Sobolev, March 30, 1955

LEVITON, M.Ye.

Some data on the structure of the Hercynian foundation of  
Ciscaucasia and adjacent regions. Sov. geol. 3 no. 11:147-  
151 N '60. (MIRA 13:12)

1. Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti.  
(Caucasus, Northern--Geology, Structural)

LEVITON, M.Ye.; MUZYCHENKO, N.M.

Basic tectonic features of the Cretaceous complex in the northwestern part of the Caspian Sea region. Dokl. AN SSSR 140 no.5:1148-1151 O '61. (MIRA 15:2)

1. Gosudarstvenny geofizicheskiy trest po razvedke nefti, gaza i ugliya. Predstavлено академиком D.V. Nalivkinym.  
(Caspian Sea Region- -Geology, Structural)

NEVOLIN, N.V.; KASATKIN, D.P.; KIREYCHEV, V.D.; KANDINOV, N.N.; LEVITON,  
M.Ye.; RTISHCHEVA, V.F.; TROITSKIY, V.N.; DYUKOV, A.I.

Structure of the recent relief of the surface basement of the  
Russian Platform. Sov.geol. 8 no.2:82-90 F '65.

(MIRA 18:12)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut geofizicheskikh  
metod razvedki.

KOVALENKO, F.; LEVITOY, A.; BEKHTIN, D.;

Simplification of the technological control apparatus at enterprises. Sots.trud no. 2:1C? -109 Ag '57. (MIRA 10:?)

1. Direktor L'govskoy obuvnoy fabriki No.3 (for Kovaleenko).
2. Inzhener zavoda "Avtosatura" (for Levitov). 3. Direktor fabriki "Vosrozhdeniye" (for Bekhtin).  
(Production control)

LEVITOV, A.

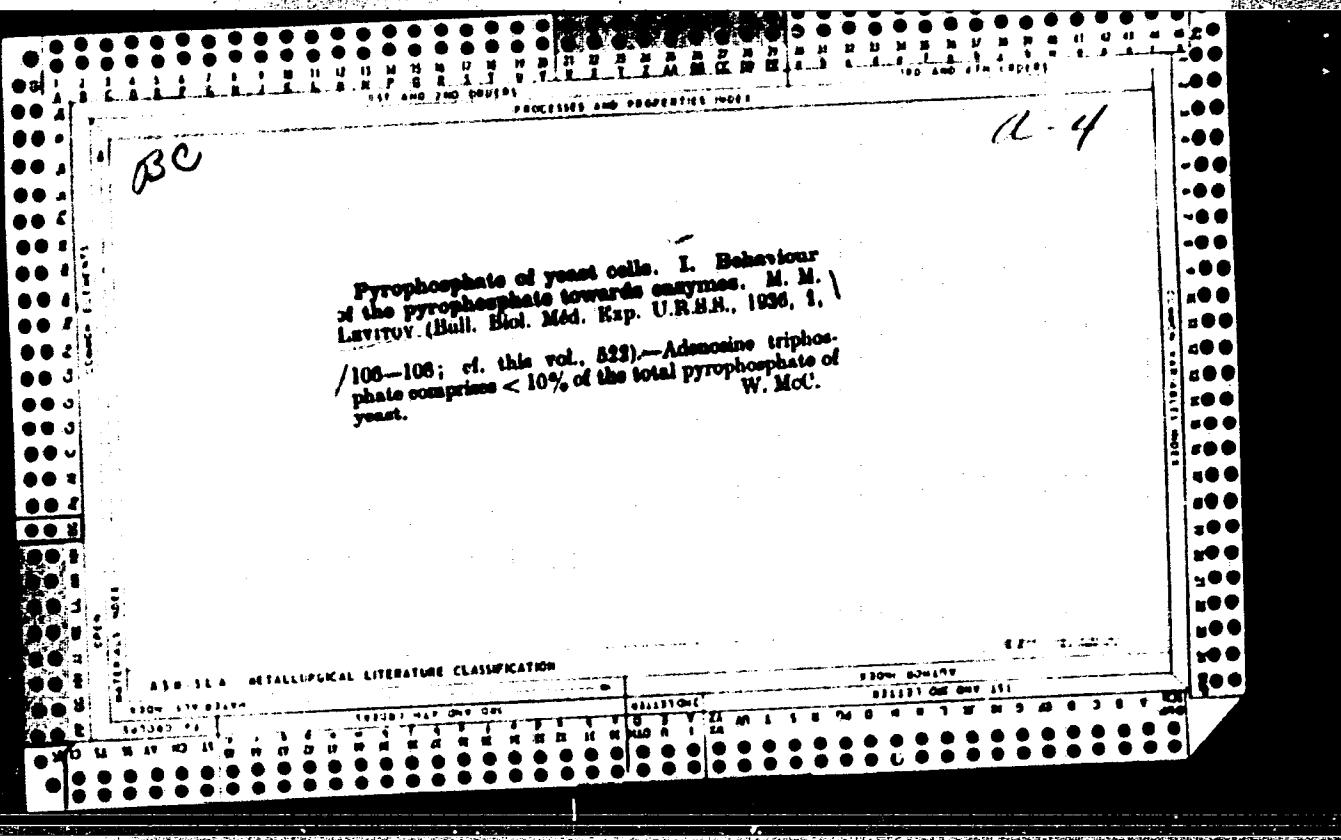
Self-financing, a tool of monopoly capital. Den. i kred. 20  
no.6:70-79 Je '62. (MIRA 15:6)  
(France—Self-financing)

LEVITOY, A.M., kandidat meditsinskikh nauk

Problem of latent brucellosis. Klin. med. 32 no.8:55-59 Ag '54.  
(MLRA 7:10)

1. Iz kafedry infektsionnykh bolezney s epidemiologiyey (zav.  
zasluzhennyy deyatel' nauki USSR prof. V.P.Petrov) Kuybyshev-  
skogo meditsinskogo instituta i Kuybyshevskoy protivo-  
brutsellesnoy stantsii.

(BRUCELLOSIS, differential diagnosis,  
latent forms)



"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITOV, M.M.

The nature of the pyrophosphate compound of yeast; M.M. LEVITOV,  
(BIOCHEMICAL DEPT. OF VIEM, MOSCOW ) vol. 1, no. 4 p. 485, 1936

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

三

Transformation of phosphorus compounds during fermentation in living bacterial cells. M. M. Levitov, N. N. Buddeev, G. G. Sos. *Biochim. Mekh.* 15(1951).—During the fermentation of glucose by *Escherichia coli*, the living, and total acid-wd. P decreases, guanidino, cyanophosphates, being formed. The amount of "alkali hydrolyzable P" is increased during fermenta-

tion of living and plasmolyzed bacteria. This P<sub>1</sub> fraction is detd. by estn. of the yeast or other cellular material with  $\text{CCl}_4/\text{CH}_3\text{OH}$ . The ppt is centrifuged, washed and subjected to hydrolysis for 10 min. with 0.05 N NaOH on a heating water bath. The hydrolysate, after cooling, is treated with  $\text{CCl}_4/\text{CH}_3\text{OH}$ , and the filtrate subjected to acid hydrolysis with  $\text{H}_2\text{SO}_4$  at 100° for 10 min. The sugar P<sub>1</sub> in this hydrolysate is detd. colorimetrically. H. B.

110

Lab. of Biochemistry of MICRObes, Viem, Moscow

BUREAU OF MINES METALLURGICAL LITERATURE CLASSIFICATION

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CIA-RDP86-00513R000929620017-5"

CA

112

Fermentation of sugar by choleraeform vibrios. M. M. Levitov and M. A. Davidovich. *Biofizika* 6, 70 (1941) (in English, no. 71 (1941); cf. "A. 34, 567"). Choleraeform vibrio (I), strain b44, consumes glucose under both aerobic and anaerobic conditions. The consumption rate is highest at pH 6-8 and is independent of coeca. (1.2-0.1%). Fermentation is fully inhibited by monodentate carboxylic acid at low concns, but only slightly by NaF. The effect of NaF is considerably weaker under aerobic than under anaerobic conditions. During fermentation inorganic phosphates are esterified. But added inorganic phosphate increases the rate of fermentation by 40-70%. It seems diphosphate is only slightly fermented by I, but its addition accelerates glucose fermentation. Glyceraldehyde has no effect on fermentation.

I. Lerner

DEPT. OF THE CHEMISTRY OF MICROBES AND BACTERIOPHAGE, VIEM, MOSCOW

A30-114 METALLURGICAL LITERATURE CLASSIFICATION

REF. NO. 2000

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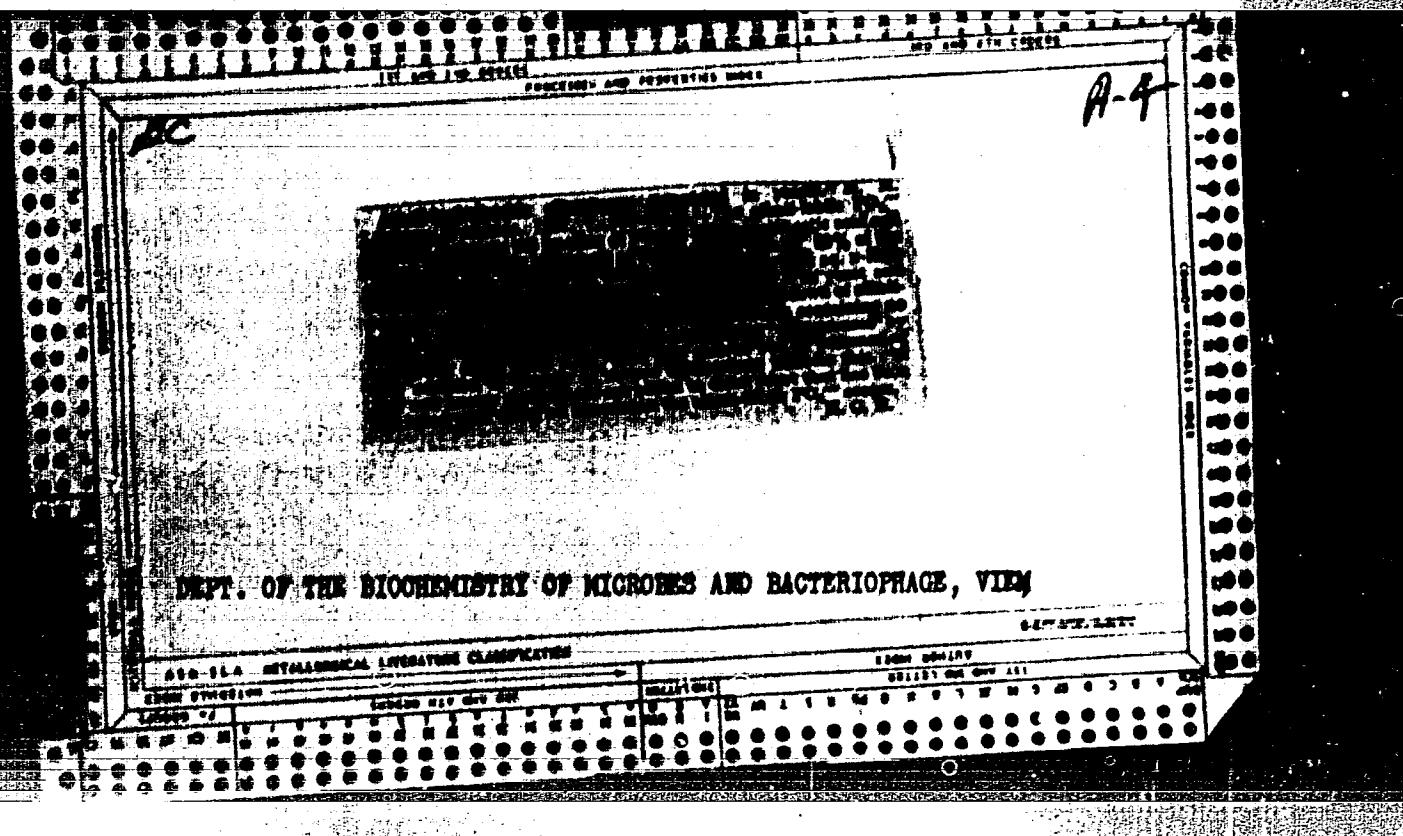
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CIA-RDP86-00513R000929620017-5"

LEVITOV, M. M.

Microbe Biochemistry Section, VI3M

"Penicylline - crustozin"

SOURCE: Zhur. Mikrobiol., Epidemiol., i Immunobiol., № 7-8, 1944

## **LEYINGE-N-N**

## STRUCTURES AND PROPERTIES OF S

**Titer and protein of antithrombin serums in plasmaapheresis.** M. N. Layton, J. G. Martindale, *Epidemiol., Immunopathol., Infect.*, 10(1964), No. 10/11, 25-7. — The method of plasmaapheresis, first described by Abel, Rowntree, and Turner (cf. *C. A.*, 6, 2231), is very useful for studying the correlation between the titer and the protein of antithrombin serums. It was found that there is no direct correlation between the antithrombin titer and the amounts of coagulation and prothrombin. In extensive bleeding plasmaaphoretic studies reveal that the greatest loss is suffered by the erythrocytes. Prothrombin is restored very quickly. By employing plasmaapheresis and reinfusion of erythrocytes, it is possible to collect large quantities of antithrombin serums without impairing their reactivity. D. I. March

116

Exp. Lab., Central Inst. Epidem. & Microbiol.

#### **510.514 METALLURICAL LITERATURE CLASSIFICATION**

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

CA

17

Pencilline. M. M. Levitov. Uspolki Khim. 13, 295-  
30X(1944).—A review covering the nature of antibiotic  
substances in general with a table showing their sources,  
phys., chem., and bactericidal properties; the discovery,  
purification, biol. and chem. properties of penicillin; the  
mechanism of its action and its medical applications.  
57 references.

F. H. Rathmann

AB-114 METALLURGICAL LITERATURE CLASSIFICATION

GA

19

Determination of penicillin. M. M. Leytus, V. D. Vyshnev, and A. M. Novashov (Inst. Biol. Prophylaxis of Infections, Moscow). *Zhikhimia* 10, 401-410 (1945) May 5-6 (English summary); cf. J. Hirsch, *Compt. rend.* **210**, 157-160 (1943-44) (publ. 1948).—In the absence of an antibiotic, the bacterial cells in broth medium continue to multiply, and as a result there is an increase of respiration, as measured by a Warburg app. When penicillin is present, the amt. of O absorbed per unit time either remains const., or increases only slightly as compared to the control. Penicillin thus acts not on respiration but on the growth of the bacteria. An 18-hr. *staphylococcus* culture is centrifuged and suspended in a meat-peptone broth with 1% glucose. In each vessel of the Warburg app. is placed the same amt. of suspension, about 40-200 million cells per ml. Penicillin is then added in various amounts to each vessel, except to the control. Peptone broth is then added so that the vol. in each vessel is the same. The temp. is kept. at 37°. Readings of the O absorbed are taken every 30 min. The method is accurate to 0.01-0.02 Oxford units, and the results are available after 1.2-2 hrs. No sterile materials are required. The method may be used not only for pure sodas. of penicillin, but also for penicillin in blood and urine. H. Prystav.

010-200 METALLURGICAL LITERATURE FLASHFEATURE

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CIA-RDP86-00513R000929620017-5"

CA

11c

**Changes in the respiratory activity of microbes grown on media with glucose.** M. M. Leviton, E. D. Vyshpan, and A. M. Nenasheva (Inst. Biol. Prophylaxis of Infections, Moscow). *Mikrobiologiya*, 11, 233-40 (1946).—As is known, the cholera vibrio is an obligatory aerobe, but if the culture medium contains glucose, the vibrio may thrive under anaerobic conditions. The same phenomenon is observed with cholera-like vibrios. These vibrios lose their respiratory power if grown on a medium contg. glucose. Their power to ferment sugar anaerobically is retained.

#### **4.16.16.4 METALLURGICAL LITERATURE CLASSIFICATION**

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

CA

11.

Directed biosynthesis of penicillin. M. M. Levitt  
(Vsesoyuz Nauch Issledovatel Inst Penitsillini i Drugih  
Antibiotikam). *Med Prom SSSR* 1949, No 4, 16-21.  
Review on the biosynthesis of the penicillins by the use of  
appropriate precursors. G. M. Kosulapoff

LEVITOV, M.M., kandidat biologicheskikh nauk; NEMASHEVA, A.N.

Effect of penicillin on gram-positive and gram-negative microbes.  
Trudy VNIIL no.1:105-113 '53. (MLRA 8:1)  
(Penicillin) (Bacteria. Pathogenic)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITOY, M.M., kandidat biologicheskikh nauk; GERMANOVA, K.I., kandidat  
meditsinskikh nauk; NEMASHEVA, A.N.

Effect of certain conditions on the antibacterial action of penicillin.  
Trudy VNIIL no.1:113-123 '53.  
(Penicillin) (MLRA 8:1)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

11) Action of eritrit on the respiration of microbes. M. M. Levitz and A. N. Bondareva. *Trudy Vsesoyuz. Nauk.-Issledovatel. Inst. Antibiotikov* 1953, No. 1, 157-64. Results obtained indicate that the antibacterial action of eritrit can be explained by its ability to affect the oxidation system of the cells. It decreased utilization of oxygen by the eritrit-sensitive *Pseudomonas* and *Naphylococcus* but it did not affect oxygen utilization in the eritrit-insensitive *Escherichia coli*. The decrease in respiration in sensitive cells was noticeable within ten minutes after addition of the antibiotics. V. Mihajlov

LEVITOV, M. M.

The influence of cultivation method upon the development of microbial resistance to streptomycin. K. I. Germanova and M. M. Levitov. *Zhur. Mikrobiol. Epidemiol. Immunobiol.* 1954, No. 2, 16-20.—Known amounts of 2 synthetic fluid media and nutrient broth were mixed with varying amounts of streptomycin and inoculated with a 24-hr. culture of *Escherichia coli*. Nutrient broth cultures required 3-5 times as much streptomycin as the synthetic media for complete suppression of bacterial growth. Those synthetic media and broth cultures which showed max. growth with the largest amt. of streptomycin were used for consecutive transplantations. After 3-7 transfers the resistance of *E. coli* grown in broth increased 200 times while that of synthetic-media grown organisms increased only 4 times after 10-12 transfers. Addn. of casein hydrolyzate to the synthetic media enhanced the resistance of the organisms to streptomycin. Addn. of nucleic acid and vitamin B complex did not affect resistance. The lower the pH the faster resistance developed in both synthetic and nutrient media. A. Shirkov

USSR/Chemistry - Antibiotics

Card 1/2      Pub. 22 - 27/54

Authors : Shemyakin, M. M., Memb.Cor.Acad. of Sc., USSR; Kolosov, M. N.; Levitov, M. M.; Gormanova, K. I.; Karapetyan, M. G.; Shvetsov, Yu. B.; Andrusas, E. M.  
Title : Relation between structure and antimicrobial activity of chloromycetin (levomycetin) and the mechanism of its reaction  
Periodical : Dok. AN SSSR 102/5, 953-956, Jun 11, 1955  
Abstract : It is shown that the high selectivity of the biological effect of chloromycetin on microbes is determined simultaneously by the following factors: 1) strong polarizing effect of the p-nitrophenyl radical, the geometrical dimensions of which are of no importance; 2) strong polarizing effect of the dichloroacetyl radical, which should satisfy even the most specific geometrical requirements; and 3) defined geometrical dimensions and corresponding conformation of the aminopropanediol group. The relation between the structure and biological activity of chloromycetin is explained.  
Institution : Acad. of Med. Sc., USSR, Inst. of Biol. and Med. Chem.  
Submitted : January 27, 1955

Card 2/2      Pub. 22 - 27/54

Periodical : Dok. AN SSSR 102/5, 953-956, Jun 11, 1955

Abstract : Five references: 2 USSR and 3 USA (1858-1955). Diagrams.

LEVITOV, M.M.; GOTOVTSVA, V.A.; INOZEMTSEVA, I.I.; BYCHKOVA, M.M.; LUR'YI,  
L.N.; KASHCHEYEV, N.A.; KENASHEVA, A.M.

Production of radioactive penicillin ( $\text{S}^{35}$ ). Antibiotiki 1 no.4:20-24  
Jl-Ag '56. (MIRA 9:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIN, radioactive  
prod.)

**EXCERPTA MEDICA Sec.4 Vol.11/4 Med.Microb. etc. 7/1953**

**797. SOME CONDITIONS OF UTILIZATION OF PRECURSORS FOR PENICILLIN SYNTHESIS BY THE FUNGUS *PENICILLIUM CHRYSEOGENUM* (Russian text) - Levitov M. M. and Tovarova I. I. Inst. of Antibiotics, Moscow - ANTIBIOTIKI 1956, 1 (21-23) Tables 2**

Studies were made on the conditions of simultaneous utilization of 2 precursors for penicillin synthesis. Growing the fungus on a synthetic medium in the presence of phenylacetamide, *para*-oxyphenylacetic acid, propylmercaptoacetic acid, or phenylseleniumacetic acid, it was found that the last 2 acids, in concentrations of 0.01-0.02%, sharply increased the activity of the culture fluid (3-4 times). Phenylacetamide was slightly less active and *para*-oxyphenylacetic acid was almost inactive. On simultaneous addition to the medium of phenylacetamide and propylmercaptoacetic acid, or phenylacetamide and phenylseleniumacetic acid the activity decreased sharply; on addition of phenylacetamide and *para*-oxyphenylacetic acid no changes at all were observed. The authors conclude that the presence in the medium of 2 penicillin precursors results in competition.

Svinkina - Moscow (S)

LEVITOV-M.M.

Chemistry of chloromycetin (levomycetin). VIII. Dependence of antimicrobial activity of chloromycetin on its structure and the mechanism of action of chloromycetin. M. M. Shemyakin, M. N. Kolosov, M. M. Levitov, K. I. Germanova, M. G. Karapetyan, Yu. N. Shvetsov and E. M. Bandas. *Zhur. Obshch. Khim.* 26, 821 (1956). cf. *C.A.* 49, 190494, 50, 3291. Biol. tests of several N-acetyl derivs of chloromycetin against *Staphylococcus aureus*, *Escherichia coli*, *Bacillus subtilis*, and *Vibrio fluorescens* were performed. The results indicate that the *p*-nitrophenyl group is important to the activity of the drug both through its electronic behavior and its polarizing action on the rest of the mol.; the geometric dimensions of this part of the mol. are not important in contrast to the import of geometric dimensions in the aminopropanediol portion of the mol. The NO<sub>2</sub> group can be shifted without loss of activity to other conjugated locations, and compds. with p-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>-N:N— or p-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>N— linkages are highly active; compds. without the NO<sub>2</sub> group or those with it in unconjugated locations (p-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>CONH—) are inactive or weakly active. The biol. activity of chloromycetin analogy drops off in the series of the *p*-phenyl substituents: NO<sub>2</sub>, CN, CO<sub>2</sub>Me, Cl, SO<sub>3</sub>Me, SO<sub>3</sub>NH<sub>2</sub>. Geometry and polarization in chloromycetin are discussed at length. New analogs were prepd. By heating 6 g. *D,L-threo-[(p-nitrophenyl)-2-aminopropyl]iso-AmOH* to 110° 5 min., followed by treatment with

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1/2

SHEMYAKIN, M.M. . .

Et<sub>2</sub>OAc, gave 17% **DL-threo-1-(p-nitrophenyl)-2-(γ,γ,γ-trichloroacetylaminoo)-1,3-propanediol**, m. 165-6° (from Cl<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>Cl). I (6 g.) in 350 ml. Et<sub>2</sub>O and 190 ml. 0.5*N* KOH treated with 8.1 g. CCl<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>COCl (b.p. 97°) 0.5 hr. gave 87% **DL-threo-1-(p-nitrophenyl)-2-(γ,γ,γ-trichlorobutyrylaminoo)-1,3-propanediol**, m. 116-17° (from (CH<sub>2</sub>Cl)<sub>2</sub>). D- or L-form of I (9 g.) similarly treated with CHCl<sub>2</sub>CH<sub>2</sub>COCl (b.p. 79-81°, n<sub>D</sub><sup>20</sup> 1.5165) gave 70.5% **D-threo-1-(p-nitrophenyl)-2-(γ,γ-dichloroacetylaminoo)-1,3-propanediol** m. 84.5° (from H<sub>2</sub>OAc and (CH<sub>2</sub>Cl)<sub>2</sub>, n<sub>D</sub><sup>20</sup> 1.5183, [α]<sub>D</sub><sup>20</sup> -67.8° (Me<sub>2</sub>CO); **L-threo analog**, m. 84-5°, [α]<sub>D</sub><sup>20</sup> +67.8° (Me<sub>2</sub>CO), b.i.-analog, prep'd by mixing the 2 isomers, m. 144-5°. I (8 g.) in 300 ml dry dioxane was treated at 12-15° with 2.45 g. CCl<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>COCl over 0.5 hr., after 0.5-hr. shaking the mixture was filtered and concd *in vacuo*, treated with Et<sub>2</sub>OAc, washed with dil H<sub>2</sub>SO<sub>4</sub> and 20% NaCl, and evapd, yielding 88% **DL-threo-1-(p-nitrophenyl)-2-(γ,γ-dichloroacetylacetamido)-1,3-propanediol II hydrate** (from heptane and (CH<sub>2</sub>Cl)<sub>2</sub> in Et<sub>2</sub>OAc-CH<sub>2</sub>Cl) m. 72.3°, the water of hydration is lost at 100° *in vacuo*. This (0.2 g.) in dry dioxane treated with 2 drops dry Et<sub>2</sub>N and kept 45 hrs gave 90% **DL-threo-1-(p-nitrophenyl)-2-(γ,γ-dichloroacetylaminoo)-1,3-propanediol**, m. 144.5° identical with the above described. Refluxing II with 20% HCl 3 hrs. gave 87% CCl<sub>3</sub>CH<sub>2</sub>CO<sub>2</sub>H and 91% I. G. M. Kosolapoff

2/2

USSR/Microbiology - Antibiosis and Symbiosis. Antibiotics.

F-2

Abs Jour : Ref Zhur - Biol., No 12, 1958, 52754

Author : Levitov, M.M.

Inst :

Title : The Biological Significance of Penicillin Formation.

Orig Pub : Antibiotiki, 1957, 2, No 2, 3-7

Abstract : Based on the author's own experiments, and literature data, a hypothesis of the biological significance of penicillin (I) is set forth. The author believes that formation of I represents a process of defensive synthesis, as a result of which substances toxic to penicillin are made harmless by combining the antibiotic with given products of mold metabolism. To the introduction of the toxic substance of the precursor (II) penicillin responds by an increased formation of I, which combines with the molecule of II. This hypothesis is confirmed by the following facts. The composition of the nucleus of the I molecule remains unchanged, and

Card 1/2

- 21 -

LEVITOY, M.M.; GERMANOVA, K.I.; TOVAROVA, I.I.; BYCHKova, M.M.; LUR'YE, L.M.;  
MIKHAYLINKOV, P.S.

Physiological characteristics of various strains of *Penicillium chrysogenum*; effect of the composition of the medium and of fermentation conditions on penicillin synthesis by strains New Type 24, Hybrid-31 and B-51-20. Antibiotiki 3 no.2:3-7 Mr-Ap '58.  
(MIRA 12:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(*PENICILLIUM*, culture,  
*chrysogenum*, eff. of medium composition & fermentation  
on penicillin synthesis by various strains (Rus))

LEVITOV, M.M.; GOTOVTSIEVA, V.A.

Biosynthesis of phenoxyethylpenicillin. Antibiotiki 3 no.5:  
21-25 S-0 '58. (MIRA 12:11)

1. Vsesoyusnyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIN, rel. cpds.  
phenoxy methyl penicillin, biosynthesis (Rus))

TOVAROVA, I.I.; LEVITOV, M.M.; GERMANOVA, K.I.

Role of precursors in the synthesis of penicillin by various strains. Antibiotiki 3 no.5:26-30 S-0 '58. (MIRA 12:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(Penicillin, rel. cpds.  
benzyl penicillin, eff. of precursors on  
synthesis by various strains (Rus))

LEVITOV, M.M.

GERMANOVA, K.I.; LEVITOV, M.M.; STEPANOVA, N.Ye.; NENASHEVA, A.M.

Physiological characteristics of various strains of Penicillium chrysogenum; certain characteristics of metabolism in strains B-51-20, 31 and 24 [with summary in English]. Antibiotiki 3 no.6:8-14 N-D '58. (MIRA 12:2)

(PENICILLIN, metabolism,  
chrysogenum B-51-20, 31 & 24 (Rus))

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITOV, M.M.; GERMANOVA, K.I.

Antibiotic research and production in the Chinese People's Republic.  
Antibiotiki, 4 no.2:118-121 Mr-Ap '59. (MIRA 12:7)  
(CHINA--ANTIBIOTICS)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

LEVITOV, M.M.; KLAPOVSKAYA, K.I.; YUDINA, O.D.

Formation of penicillin nucleus during fermentation and its conversion to penicillin. Antibiotiki 4 no.6:18-24 N-D '59.  
(MIRA 13:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIN chem.)

LYO ZHO-IN [Liu Jo-ying]; TUN SYU-IN [T'ung Hsiau-ying]; TSEN SYUO-KHEN;  
LEVITOV, M.M.

Influence of lactic acid on the synthesis of oxytetracycline in fer-  
mentation. Antibiotiki 5 no.1:52-57 Ja-F '60. (MIRA 13:7)

1. Laboratoriya biokhimii Instituta antibiotikov Kitayskoy Narodnoy  
Respubliki.  
(OXYTETRACYCLINE) (LACTIC ACID)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITOV, M. M., GOTOVTSEVA, V. A., YUDINA, O. M., VERKHOVTEVA, T. P.,  
and KLAPOVSKAYA, K. I. (USSR)

"The Biosynthesis of Penicillins and Penicillin-like Substances in  
fermentation without a precursor."

Report presented at the 5th International Biochemistry Congress,  
Moscow, 10-16 Aug 1961

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

SHOSTAKOVSKIY, M.F.; RABINOVICH, M.S.; LEVITOY, M.M.; VVERKHOTSEVA, T.P.;  
PREOBRAZHENSKAYA, Ye.V.; KULIKOVA, G.N.; KALINOVSKIY, O.A.

Synthesis of the precursors and fragments of antibiotics. Part 4:  
Thioglycolic acid derivatives. Zhur.ob.khim. 31 no.5:1453-1458  
(MIRA 14:5)  
My '61.

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(Acetic acid) (Antibiotics)

LEVITOV, M.M.; VVERKHVTSEVA, T.P.; RABINOVICH, M.S.; PREOBRAZHENSKAYA, Ye.V.;  
KULIKOVA, G.N.; BUIANOVSKAYA, I.S.; SIMEYENSON, A.N.

Biosynthesis of new penicillins using propylmercaptoacetic  
acid derivatives as precursors. Antibiotiki 6 no.7:575-581  
(MIRA 15:6)  
Jl '61.

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIN) (ACETIC ACID)

BURGEL'SON, I.D.; LEVITOV, M.M.; MOLOTKOVSKIY, Yul.G.; SAZYKIN, Yu.S.;  
SHEMYAKIN, M.M.

Synthesis and study of the antimicrobial action of the simplest  
analogues of macrolide antibiotics. Antibiotiki 6 no.7:581-585  
(MIRA 15:6)  
Jl '61.

1. Institut khimii prirodnnykh soyedineniy AN SSSR.  
(ANTIBIOTICS)

POPOVA, L.A.; LEVITOV, M.M.; BELOZEROVA, O.P.

Effect of fats on the biosynthesis of chlortetracycline.  
Antibiotiki 6 no.11:989-994 N '61. (MIRA 15:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(AUREOMYCIN) (OILS AND FATS)

LEVITOV, M.M.; LUR'YE, L.M.; ZAVILEYSKAYA, G.F.

Role of precursors in the biosynthesis of penicillin. Antibiotiki 6  
no.12:1058-1063 D '61. (MIRA 15:2)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIN)

LEVITOV, M.M.; INOZEMTSEVA, I.I.; GOTOVTSVA, V.A.; KOMOKINA, Z.F.;  
YUDINA, O.D.; KLEINER, G.I.; IOFFE, R.I.; NAGLE, A.M.

Production and basic properties of almecillin (allylmercaptomethyl-  
penicillin). Med. prom. 15 no.11:12-19 N '61. (MIRA 15:6)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov  
i Rizhskiy zavod meditsinskikh preparatov.  
(PENICILLIN)

LEVITOV, M.M.; INOZEMTSEVA, I.I.; TEBYAKINA, A.Ye.; BUYANOVSKAYA, I.S.;  
SHNEYERSON, A.H.; CHAYKOVSKAYA, S.M.; KOKOKINA, Z.F.; DRUZHININA, Ye.N.

New type of penicillin —  $\alpha$ -phenoxyethylpenicillin and study of  
its microbiological properties. Antibiotiki 7 no.2:104-108 F '62.  
(MIA 15:2)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIN)

LEVITOV, M.M.; YUDINA, O.D.

Study of the respiration of *Penicillium chrysogenum*. Antibiotiki  
7 no.3:25-30 Mr '62. (MIRA 15:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov.  
(PENICILLIUM)

"APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5

LEVITOV, M.M.; SAVITSKAYA, Ye.M.; TOVAROVA, I.I.

6-Aminopenicillanic acid and new penicillins. Antibiotiki 7 no.5:  
387-398 My '62. (MIRA 15:4)  
(PENICILLANIC ACID) (PENICILLIN)

APPROVED FOR RELEASE: 07/12/2001

CIA-RDP86-00513R000929620017-5"

LEVITOV, M.M.; GOTOVTSEVA, V.A.; ZAVILEYSKAYA, G.F.

Formation of 6-aminopenicillanic acid during the fermentation  
of *Penicillium chrysogenum* on a medium without a precursor.  
Antibiotiki 7 no.5:410-414 My '62.

(MIRA 15:4)

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